

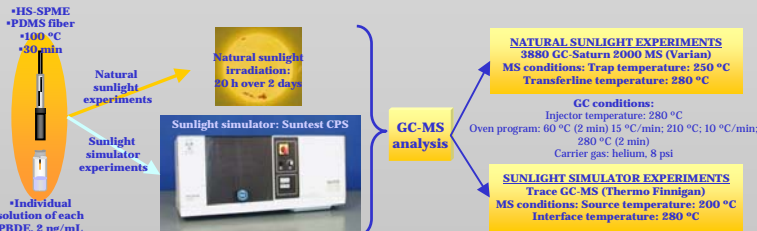
Lucía Sánchez-Prado¹, Josep Maria Bayona², Marta Lores, Maria Llompart, Carmen García-Jares, Rafael Cela
Departamento de Química Analítica, Nutrición y Bromatología, Facultad de Química, Instituto de Investigación y Análisis Alimentario, Universidad de Santiago de Compostela, Avda de las Ciencias s/n, E-15782 Santiago de Compostela, Spain.

²Departamento de Química Medioambiental I.I.Q.A.B.-C.I.D.-C.S.I.C. Jordi Girona, 18-26, E-08034 Barcelona, Spain.
*lspko@usc.es

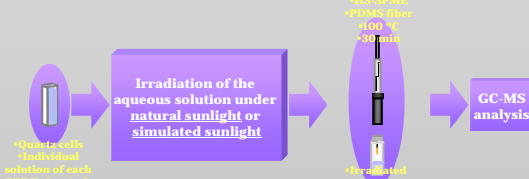
INTRODUCTION

- Sunlight photolysis is one of the most important potential abiotic degradation pathways for polybrominated diphenyl ethers (PBDEs) [1].
- Previous work of our research group using the photo solid phase microextraction approach (photo-SPME) [2, 3] have contributed to extend the knowledge about the kinetics and the identities of the photoproducts produced during UV photochemical decomposition of these compounds.
- In photo-SPME, the SPME fiber is employed as a support for photochemical studies. First the target analytes are extracted by SPME, and then the fiber is exposed to the radiation for the required time. Thus, the sunlight photolytical behaviour of five PBDEs (BDE 47, BDE 100, BDE 99, BDE 154 and BDE 153) has been studied.
- Aqueous photodegradation studies have also been carried out. In this case, SPME is only used as extraction procedure.
- Up to now, photo-SPME studies had been carried out using UV irradiation at 254 nm [4, 5]. In the present study, the photo-SPME approach is combined for first time with both natural sunlight irradiation and from a solar simulator (Suntest CPS, Atlas).

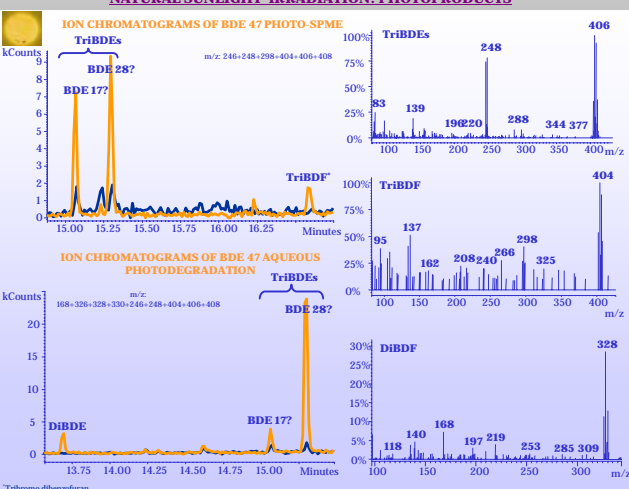
EXPERIMENTAL PHOTO-SPME PROCEDURE



EXPERIMENTAL AQUEOUS PHOTODEGRADATION PROCEDURE



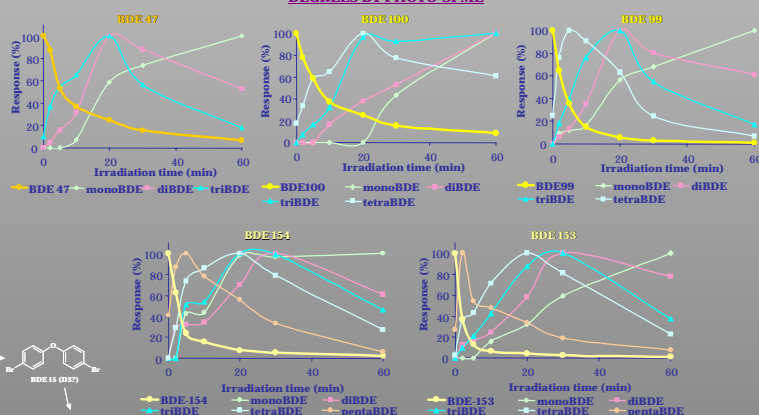
NATURAL SUNLIGHT IRRADIATION: PHOTOPRODUCTS



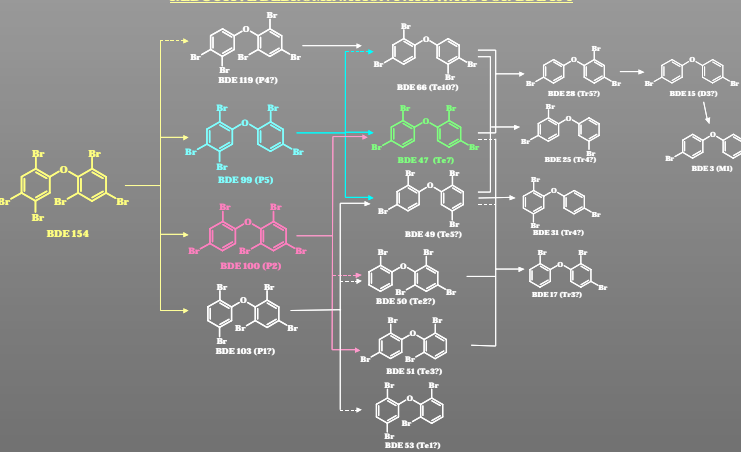
KINETIC PARAMETERS IN THE SOLAR SIMULATED EXPERIMENTS: PHOTO-SPME vs AQUEOUS PHOTODEGRADATION

	Photo-SPME		Aqueous photodegradation	
	k (min ⁻¹)	t _{1/2}	k (min ⁻¹)	t _{1/2}
BDE 47	0.103	6.72	0.167	4.16
BDE 100	0.096	7.19	0.150	4.62
BDE 99	0.187	3.71	0.317	2.18
BDE 154	0.189	3.67	0.288	2.41
BDE 153	0.401	1.73	0.334	2.07

PHOTOFORMATION AND DEGRADATION OF PBDEs OF DIFFERENT BROMINATION DEGREES BY PHOTO-SPME



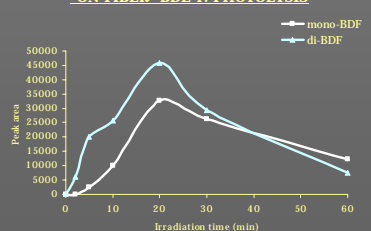
REDUCTIVE DEBROMINATION PATHWAYS FOR BDE 154



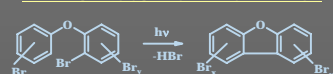
BDEs SIMULATED SUNLIGHT IRRADIATION PRODUCTS OBTAINED IN THE PHOTO-SPME EXPERIMENTS

Congeners	Key	Proposed identification	RT (min)	RT range (min)	Identification ions	Source			
monoBDE	M1	BDE 3	12.07	12.07	248+250	All studied BDEs			
	diBDE	D1	BDE 8	14.02	14.02-14.32	168	BDE47, BDE100, BDE99, BDE153		
		D2	BDE 13	14.15		326+328+330	BDE99, BDE153		
		D3	BDE15	14.32		All studied BDEs			
		Tr1	BDE 18	15.45		BDE99			
		Tr2	BDE 32	15.57		BDE100, BDE153			
		Tr3	BDE 17	15.75		All studied BDEs			
		triBDE	Tr4	BDE 25 or BDE 31		15.82	15.45-16.42	246+248	BDE99, BDE154, BDE153
			Tr5	BDE 28		16.02		404+406+408	BDE99, BDE154, BDE153
			Tr6	BDE 37		16.42		All studied BDEs	
Te1			BDE 53	17.13		BDE99, BDE153			
Te2	BDE 50		17.20	BDE154					
Te3	BDE 51		17.55	BDE100, BDE154					
Te4	BDE 75		17.60	BDE100					
tetraBDEs	Te5		BDE 49	17.72	17.13-18.48	324+326+328		BDE99, BDE154, BDE153	
	Te6		BDE 48	17.77		484+486+488		BDE99, BDE153	
	Te7		BDE 47	18.10		BDE100, BDE99, BDE154, BDE153			
	Te8	BDE 52 or BDE 70	18.20	BDE153					
	Te9	BDE 74	18.28	BDE99, BDE153					
	Te10	BDE 66	18.48	BDE99, BDE154, BDE153					
	P1	BDE 103	19.22	BDE154					
	P2	BDE 100	19.97	BDE154					
	pentaBDEs	P3	BDE 101	20.13		19.22-21.23	404+406+408	BDE153	
		P4	BDE 119	20.23			562+564+566	BDE154	
P5		BDE 99	20.62	BDE154, BDE153					
P6		BDE 118	21.23	BDE153					

TIME PROFILES OF THE PBDFs GENERATED UPON "ON-FIBER" BDE 47 PHOTOLYSIS



FORMATION PATHWAYS OF PBDEs FROM PBDE DURING SUNLIGHT DEGRADATION



CONCLUSIONS

The photo-SPME technique can be successfully combined with sunlight irradiation, either natural or artificial. The photochemical behaviour in the fiber mimics the processes that happen in water also under sun irradiation (verified here for PBDEs as a case study). The results obtained under simulated sunlight match with the results obtained in the previous work under UV irradiation. Photodegradation pathways for the five BDEs under consideration are for first time proposed for simulated sunlight irradiation. Reductive debromination is the main pathway. PBDEs have been formed under both natural and solar simulated irradiation.

REFERENCES

- [1] W.U. Palm, R. Koptzyk, W. Sossinka, W. Ruck, C. Zetzsch. *Organohal. Comp.* 66 (2004) 4105-4110. [2] L. Sánchez-Prado, M. Llompart, M. Lores, C. García-Jares, R. Cela, J. Chromatogr. A. 1071 (2005) 85-92. [3] L. Sánchez-Prado, C. González-Barreiro, M. Lores, M. Llompart, C. García-Jares, R. Cela. *Chemosphere.* 60 (2005) 922-928. [4] M. Lores, M. Llompart, R. González-García, C. González-Barreiro, R. Cela. *J. Chromatogr. A.* 963 (2002) 37-47. [5] M. Lores, L. Sánchez-Prado, M. Llompart, C. García-Jares, R. Cela. *Intern. J. Environ. Anal. Chem.* 85 (2005) 281-291.

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